

BJP

**Bangladesh Journal of Pharmacology** 

Research Article

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A Journal of the Bangladesh Pharmacological Society (BDPS)

Journal homepage: www.banglajol.info; www.bdpsjournal.org

Abstracted/indexed in Academic Search Complete, Agroforestry Abstracts, Asia Journals Online, Bangladesh Journals Online, Biological Abstracts,

BIOSIS Previews, CAB Abstracts, Current Abstracts, Directory of Open Access Journals, EMBASE/Excerpta Medica, Global Health, Google Scholar,

HINARI (WHO), International Pharmaceutical Abstracts, Open J-gate, Science Citation Index Expanded, SCOPUS and Social Sciences Citation Index

ENSN: 1991-0088

## Synthesis, in silico analysis and antidepressant activity of 1,3,4oxadiazole derivatives

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#### **Article Info**

Received: 15 March 2022 Accepted: 31 March 2022 31 March 2022 Available Online: DOI: 10.3329/bjp.v17i1.58728

Cite this article: Gatphoh BFD, Aggarwal NN, Kumar SM, Kumar MV, Revanasiddappa BC. Synthesis, in silico analysis and antidepressant activity of 1,3,4-oxadiazole derivatives. Bangladesh J Pharmacol. 2022; 17: 14-21.

## **Abstract**

The compounds 1,3,4-oxadiazole derivatives (1-8) were synthesized by the cyclization of 4-hydroxy benzhydrazide (1) with various substituted aromatic aldehydes (2) using FeCl<sub>3</sub> as catalyst and methanol as a solvent medium. The structures of the newly synthesized compounds were assigned based on FT-IR, <sup>1</sup>H-NMR, and mass spectral data. In vivo antidepressant activity was performed by tail suspension test and forced swimming test models. Using the Schrodinger Maestro, the in silico analysis was performed and docked to the glycogen synthase kinase 3β binding site (PDB: 3GB2). Compounds 8 [4,4'-(1,3,4-oxadiazole-2,5-diyl)diphenol] and 3 [3-(5-(4-hydroxyphenyl)-1,3,4 -oxadiazol-2-yl) phenol] showed both potent inhibitory activity against GSK-3β with a docking score of -7.800 kcal/mol as well as good antidepressant activity in both tail suspension and forced swimming tests models. The synthesized derivatives showed good antidepressive potential.

## Introduction

1,3,4-Oxadiazoles are an important class of compounds in medicinal chemistry because of their interesting diversified biological applications (Arshad et al., 2014). The activities include antibacterial (Maslat et al., 2002), cytotoxic (Puthiyapurayil et al., 2012), anti-cancer (Aboraia et al., 2006), antitubercular (Macaev et al., 2011), anti-inflammatory (Omar et al., 2006), antidepressant (Tantray et al., 2018), anticonvulsant (Omar et al., 2006), analgesic (Bhandari et al., 2008), hypolipidemic (Patel et al., 2014), anti-fungal (Chen et al., 2007), etc,. There are some examples of compounds currently used in clinical practice containing 1,3,4-oxadiazole moiety, such as raltegravir (used to treat HIV infection), nesapidil (antiarrhythmic), furamizole (antibacterial), tiodazocin (antihypertensive) and, zibotentan (anticancer).

Depression is a life-threatening psychological illness and is linked to impaired, poor quality of life, raised health-related costs, and is considered a major risk factor for many ailments, including cardiovascular, metabolic, and psychiatric disorders. (Lou, 2003; Otte, 2016; Casey et al., 2005). Current drug therapy approaches have limited effectiveness and are associated with many adverse health effects. Consequently, a better awareness of this depression's pathophysiology alongside the development of innovative and advanced therapies contributes significantly (Predictable, 2006; Tamminga et al., 2002).

Glycogen synthase kinase 3 (GSK-3), discovered in the late 1970s, occurs in two isoforms: GSK-3 alpha and GSK-3 beta, where  $\beta$  isozyme predominates the human brain. These receptors, a proline-directed serine-threonine kinase, are constitutively an active protein kinase



requiring primed phosphorylation of most of its substrates (Beurel et al., 2015) increases activation of GSK- $3\beta$  and has been found in the prefrontal cortices of postmortem depressed suicide victim (Karege et al., 2012). Glycogen synthase kinase  $3\beta$  (GSK- $3\beta$ ) inhibition is anticipated to be a desired pharmacological approach for treating depression. Lithium is used to treat bipolar disorder was reported to inhibit around 25% of total GSK-3 activity (Kramer et al., 2012). Both fluoxetine, which is a specific serotonin reuptake inhibitor (SSRI), and Imipramine, a tricyclic antidepressant (TCA), have been shown to increase the phospho-Ser9-GSK- $3\beta$  level, which otherwise reduces the effects of antidepressants when phosphorylation is impeded (Li et al., 2004).

This inhibition has been achieved by the high binding affinity of Imipramine to 5-HT<sub>2A</sub> receptor, SERT (serotonin reuptake transporter), and norepinephrine transporter (NET). Norepinephrine acting through protein kinase A and C, and 5-HT acting through receptors other than 5-HT<sub>2A</sub> receptor, inactivate GSK3- $\beta$ , whereas the blockade of 5-HT<sub>2A</sub> receptor prevents the inhibition of AKT. Similarly, fluoxetine inhibits SERT and increases the concentrations of 5-HT in the synaptic cleft (Duda et al., 2020).

Further, 2,5-disubstituted-1,3,4-oxadiazoles are applied to various biological activities due to the -N=C-O- bonding in the ring (Desai, 2016; Roy, 2017), Considering the broad range of biological attributes, it was contemplated to synthesize a new series of 1,3,4-oxadiazole derivatives and evaluated for their *in vivo* antidepressant activity in mice. The compounds have been docked in the groove binding side of the GSK-3 $\beta$  receptor to verify their binding affinity and inhibition effect.

## **Materials and Methods**

The required chemicals and solvents were procured from Sigma Aldrich, India and Himedia India. Melting points were measured by digital melting point equipment from Equiptronics and were uncorrected. Thin layer chromatography plates, precoated with silica gel G, were used to monitor the completion of the reaction (solvent system: ethyl acetate: methanol (2:1 v/v). IR spectra were measured using KBr disc on Alpha Bruker FT-IR infrared spectrophotometer (cm-1). Agilent 400MR DD<sub>2</sub> spectrometer was used to generate  $^1\text{H-NMR}$  spectra with d<sub>6</sub>-DMSO/CDCl<sub>3</sub> solvent and tetramethylsilane as an internal standard. In  $\delta$  ppm, the chemical shift values were recorded, and the mass spectrum was acquired using a Waters LC-MS/ MS.

# General procedure for the synthesis of 1,3,4-oxadia-zoles (1-8)

4-Hydroxybenzhydrazide (0.01 mol) (1) and substituted aromatic aldehydes (2) (0.01 mol) were dissolved in 30

ml of methanol and a of pinch of FeCl<sub>3</sub> is added and the reaction contents were refluxed for 6-8 hours. After refluxing, the reaction contents were poured into crushed ice and stirred vigorously. The crude product obtained was washed with water and recrystallized from ethanol.

## In silico analysis

The molecular modeling studies (Lipinski's RO5, molecular docking, ADMET properties) were carried out using Schrodinger 2018-3 suite device Maestro 11.7.012, (Ligprep, Glide XP docking, QikProp). The threedimensional crystal structure of Human GSK-3β inhibitor complex with 2-methyl-5-(3-{4-[(S)-methyl-sulfinyl] phenyl}-1-benzofuran-5-yl)-1,3,4-oxadiazole (PDB ID: 3GB2) was downloaded from the data bank of protein molecules database (http://www.rcsb.org) with a recorded resolution of 2.40 Å (Saitoh et al., 2009). In the Schrödinger suite of tools, the structures of the ligands were subjected to Maestro's LigPrep module before the molecular docking, and their structures were transformed from 2D to 3D by including ionization, stereochemical, tautomeric variations, including energy minimization and optimized for their geometry, desalinated and corrected for their chiralities and missing atoms of hydrogen. The protein 3GB2 was prepared using a protein preparation wizard, and the hydrogen atoms were optimized throughout the protein structure. At the groove binding site of the 3GB2 protein, the synthesized derivatives of 1,3,4oxadiazoles were docked, and the docking process was performed using extra precision (XP) mode. Using the Molinspiration (http:/www. molinspiration.com/cgibin/properties), bioactivity scores for G-protein coupled receptor (GPCR) ligand, ion channel modulator, kinase inhibitor, nuclear receptor ligand, protease inhibitor and enzyme inhibitor were measured. The toxicity predictions of the synthesized derivatives were carried out using lazy structure-activity relationships (LAZAR) software that produces local QSAR (quantitative structure-activity relationship) models for each predictable compound.

#### Antidepressant activity

#### Animals

The animals were procured from Nitte University Center for Animal Research and Experimentation, Mangalore, India. Swiss albino mice weighing around 20–25 g have been collected. Animals were acclimatized, two weeks before the start of the study. Housing conditions were maintained at 25  $\pm$  2°C for 12 hours day/night cycles, and the animals were supplied with regular commercial pellet feed and water.

#### Acute toxicity studies

Standard husbandry conditions were maintained for the acute toxicity studies of 1,3,4-oxadiazole derivatives

## **Box 1: Tail Suspension Test**

#### Principle

In the tail suspension test, antidepressant effect of an experimental drug in investigated.

## Requirements

Synthesized 1,3,4-oxadiazole derivatives (100 mg/kg; 0.5 mL/kg); Imipramine (standard drug; 10 mg/kg); tween 80 (1%);

#### Procedure

Step 1: The mice were assigned to different groups on the day of the experiment (n=6 for each group), and the test substance was administered orally 60 min prior to the test in a fixed rotation and was placed back in their home cages.

(OECD, 2001) on female albino mice (20-25 g). Before the experimentation, the animals were fasted overnight and were treated with a single dose of 1,3,4-oxadiazoles derivatives and monitored for 48 hours for mortality (short-term toxicity). The next doses were calculated based on the short-term toxicity profile according to OECD guidelines No. 425 (Chemicals et al., 2003)

## Forced swimming test

In forced swimming test, mice are forced to swim in a small cylinder where they are unable to escape. The suspension of synthesized 1,3,4-oxadiazole derivatives (100 mg/kg) and standard drug in 1% aqueous solution of tween 80 was administered (p. o.) to mice at a volume of 0.5 mL/body weight, where control was maintained, and Imipramine (10 mg/kg) was employed as the standard drug. The mice were assigned to different groups on the day of the experiment (n=6 for each group), and the test substance was administered 60 min before the test in a fixed rotation and was placed back in their home cages. In an open cylindrical container (25 cm in height), 10 cm in diameter containing water to a height of 18 cm and observed for 6 min, the mice were individually forced to swim. Animals were allowed to swim for 6 min, and during the last 4 min, the period of immobility was observed. In the first 2 min, the animals will demonstrate initial vigorous struggle. During the last 4 min, the immobility time was recorded (Can et al., 2012). The percentage change from control was calculated using the formula below.

% Change of immobility =  $[(test/control) \times 100) - 100]$ 

## Statistical analysis

The findings are described as mean  $\pm$  SEM, n=6.For a statistical demonstration followed by Dunnett's multiple comparison test, a one-way ANOVA was used, and p<0.05 was deemed significant. The above data is presented as mean  $\pm$  SEM, n=6.

## **Results**

#### Chemistry

The reaction sequence leading to the formation of title

Step 2: The mice were then hung, using adhesive tape, at the height of 65 cm from the floor, and 1 cm from the tail tip.

Step 3: The mice were allowed to hang for 6 min, and during the last 4 min, the time of immobility was observed.

#### Calculation

The percentage change from control was calculated using the formula below.

% Change of immobility =  $[(test/control) \times 100) - 100]$ 

#### References

Steru et al., 1985; Aslam, 2016

#### References (Video)

Can et al., 2012; Haq et al., 2019

compounds is shown in Scheme 1. The physical data of the compounds (1-8) is given in Table I.

4-(5-(4-chlorophenyl)-1,3,4-oxadiazol-2-yl)phenol (1): M.P.: 260-262 °C, Yield: 82.08%. FT-IR (KBr, v, cm<sup>-1</sup>): 3239 (OH), 2925 (C-H), 1600 (C=N), 1549 (C=C), 1076 (C -O-C), 746 (C-Cl).  $^1$ H-NMR (DMSO-d<sub>6</sub>, 400 MHz) δ = 6.91-8.41 (m, Ar-H, 8H), 11.86 (s, OH, 1H). MS (m/z): 273.69 (M+1). 82.08%

**4-(5-(4-methoxyphenyl)-1,3,4-oxadiazol-2-yl)phenol** (2): M.P.: 246-248 °C. Yield: 75.05% FT-IR (KBr, v, cm<sup>-1</sup>): 3277 (OH), 1603 (C=N), 1550 (C=C),2926 (C-H),1030 (C-O-C). <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>, 400 MHz) δ = 3.78 (s, OCH<sub>3</sub>, 3H), 6.93-8.38 (m, Ar-H, 8H), 11.74 (s, OH, 1H). MS (m/z): 269.27 (M+1).

**3-(5-(4-hydroxyphenyl)-1,3,4-oxadiazol-2-yl)phenol (3):** M.P.: 218-220°C. Yield: 78.46% FT-IR (KBr, v, cm<sup>-1</sup>): 3237 (OH), 2962 (C-H), 1606 (C=N), 1521 (C=C), 1080 (C-O-C).  $^1$ H-NMR (DMSO-d<sub>6</sub>, 400 MHz)  $\delta$  = 6.93-8.38(m, Ar-H, 8H), 11.75 (s, OH, 1H), 11.86 (s, OH, 1H). MS (m/z): 255.24 (M+1). 78.46%

**4-(5-(3-nitrophenyl)-1,3,4-oxadiazol-2-yl) phenol (4):** M.P.: 182-184°C. Yield: 79.87%. FT-IR (KBr, v, cm<sup>-1</sup>): 3226 (OH), 2928 (C-H), 1606 (C=N), 1552 (C=C), 1032 (C-O-C). <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>, TMS, 400 MHz) δ 6.93-8.55 (m, Ar-H, 8H), 11.67(s,OH,1H). MS (m/z):284.35(M+1).

**4-(5-(4-(dimethyl amino) phenyl)-1,3,4-oxadiazol-2-yl) phenol (5):** M.P.: 236-238°C. Yield: 77.37%. FT-IR (KBr, v, cm<sup>-1</sup>): 3276 (OH), 2962 (C-H), 1626 (C=N), 1555 (C=C), 1087 (C-O-C). <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>, TMS, 400 MHz) δ 2.55 (s, 2XCH<sub>3</sub>, 6H), 6.72-8.29 (m, Ar-H, 8H), 11.60(s, OH, 1H). MS (m/z): 282.33 (M+1).

**4-(5-(2-chlorophenyl)-1,3,4-oxadiazol-2-yl)phenol (6):** M.P.: 255-257°C. Yield: 81.88%. FT-IR (KBr, v, cm<sup>-1</sup>): 3214 (OH), 2928 (C-H), 1600 (C=N), 1545 (C=C), 1087 (C-O-C), 752 (C-Cl). <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>, TMS, 400 MHz) δ ppm= 6.92-8.84 (m, Ar-H, 8H), 11.77 (s, OH, 1H). MS (m/z): 273.32 (M+1).

**4-(5-phenyl-1,3,4-oxadiazol-2-yl) phenol (7):** M.P.: 205-207°C. Yield: 76.33%. FT-IR (KBr, v, cm<sup>-1</sup>): 3214 (OH), 2933 (C-H), 1606 (C=N), 1558 (C=C), 1063 (C-O-C). <sup>1</sup>H-

NMR (DMSO-d<sub>6</sub>, TMS, 400 MHz)  $\delta$  6.92-8.65 (m, Ar-H, 8H), 11.82 (s, OH, 1H). MS (m/z): 239.30 (M+1).

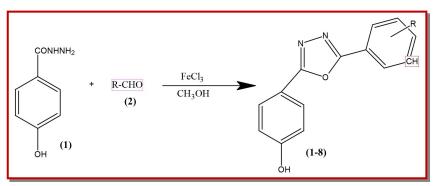
**4,4'-(1,3,4-oxadiazole-2,5-diyl) diphenol (8):** M.P.:223-225°C. Yield: 75.15%. FT-IR (KBr, v, cm<sup>-1</sup>): 3265 (OH), 2975 (C-H), 1598 (C=N), 1555 (C=C), 1063 (C-O-C). <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>, TMS, 400 MHz) δ 6.81-8.33 (m, Ar-H, 8H), 11.67 (s, OH, 1H), 11.95 (s, OH, 1H). MS (m/z): 255.30 (M+1).

The final synthesized compounds were characterized by spectral analysis (FT-IR,¹H NMR, and LC-MS/MS). A good yield ranging from 75-82% was obtained from the synthesized derivatives. The IR spectrum of compound 8 showed an absorption band at 2975 cm-1 due to aromatic C-H stretch. The absorption bands at 1598 and 1555 cm<sup>-1</sup> were attributed due to the C=N stretching and C=C stretching, respectively. The presence of OH group was observed at 3265 cm-1. The absence of carbonyl band supported the formation of 1,3,4-oxadiazole system. The <sup>1</sup>H-NMR spectrum of compound 8 showed the aromatic protons resonated as multiplets at  $\delta$  6.81-8.33. The presence of OH group was observed at δ 11.67 and 11.95 respectively integrating for one proton. Further evidence for the formation of 1,3,4-oxadiazoles (8) was obtained by recording its mass spectrum. The mass spectrum of the compound (8) showed a molecular ion peak at m/z 255.30 (M+1) in conformity with the molecular formula  $C_{14}H_{10}N_2O_3$ .

#### In silico analysis

All the newly synthesized compounds possessed desired physicochemical properties with no violations from the normal ranges and are tabulated in Table I. For all design derivatives, the number of rotatable bonds is less than 10. The tPSA values of all the compounds are within the limit indicating the cell permeability. The synthesized compounds obey Lipinski's rule of five (Table II). The compounds possess hydrogen bond donors ranging from 1(1) - 2(8) and hydrogen bond acceptors ranging from 2.250 (1) -4.250 (5). The affinity of these derivatives (1-8) with GSK-3 $\beta$  receptor (PDB ID:3GB2) was expressed in the G score.

The synthesized derivatives were found to have binding free energy in the range of -7.800 to -6.398 kcal/mol. The active residues in 3GB2 are ILE 62, GLY 63, ASN 64, GLY 65, SER 66, PHE 67, LYS 65, ASP 200, GLY 202, ASP 181, LYS 183, GLN 185, ASN 186, THR 138, TYR 140, ARG 141, ARG 144, VAL 61,VAL 70, LYS 85. Compound 8 displays the highest affinity and binding energy of -7.800 in comparison to the other synthesized derivatives. Compound 8 and 3 fit into the binding cleft of 3GB2 receptor with the binding energy of -7.800



Scheme 1: Reaction sequence leading to the formation of 8 compounds

Table I									
Physicochemical properties of compounds (1-8)									
Compound	R-CHO	Molecular Formula	MR [cm³/mol]	tPSA (<140) [Å]	Polariz ability [ų]	Nrobs (<10)	Volume (<2000) [cm³]		
1	4-C1	C <sub>14</sub> H <sub>9</sub> Cl N <sub>2</sub> O <sub>2</sub>	$70.7 \pm 0.3$	59.2	$28.0 \pm 0.5 \ 10^{-24}$	2	221.7		
2	4-OCH <sub>3</sub>	$C_{15}H_{12}N_2O_3$	$72.5 \pm 0.3$	68.4	$28.7 \pm 0.5 \ 10^{-24}$	3	233.7		
3	3-OH	$C_{14}H_{10}N_2O_3$	$67.7 \pm 0.3$	79.4	$26.8 \pm 0.5 \ 10^{-24}$	2	216.2		
4	$3-NO_2$	$C_{14}H_{9}N_{3}O_{4}$	$72.3 \pm 0.3$	105.0	$28.7 \pm 0.5 \ 10^{-24}$	3	231.5		
5	4-N(CH <sub>3</sub> ) <sub>2</sub>	$C_{16}H_{15}N_3O_2$	$80.1 \pm 0.3$	62.4	$31.8 \pm 0.5 \ 10^{-24}$	3	254.0		
6	2-C1	C <sub>14</sub> H <sub>9</sub> Cl N <sub>2</sub> O <sub>2</sub>	$70.7 \pm 0.3$	59.2	$28.0 \pm 0.5 \ 10^{-24}$	2	221.7		
7	$C_6H_5$	$C_{14}H_{10}N_2O_2$	$65.8 \pm 0.3$	59.2	$26.1 \pm 0.5 \ 10^{-24}$	2	208.1		
8	4-OH	$C_{14}H_{10}N_2O_3\\$	$67.7 \pm 0.3$	79.4	$26.8 \pm 0.5  10^{-24}$	2	216.2		
Standard (Imipramine)			$88.9 \pm 0.3$	6.5	$35.3 \pm 0.5  10^{-24}$	1	269.2		

Table II								
Lipinski's rules and dock scores at active site of 3GB2 (1-8)								
Compound	Molecular weight	Log P	Donor_HB	Acceptor HB	Dock score			
1	272.7	4.55	1	2.250	-7.132			
2	268.3	2.75	1	4.000	-6.560			
3	254.3	3.43	2	4.000	-7.524			
4	283.2	3.21	1	4.250	-6.529			
5	281.3	3.16	1	4.250	-6.963			
6	272.7	3.44	1	3.250	-6.465			
7	238.3	2.88	1	3.250	-6.398			
8	254.3	2.49	2	4.000	-7.800			
Standard (Imipramine)	280.4	4.32	0	4.00	-9.08			

kcal/mol and -7.524 kcal/mol, respectively. The hydrogen bond interactions are formed with ASP 200, LYS 183 and ARG 144 (Figure 1) in the case of 8 and ILE 62, and ASP 200 in the case of 3. The hydrophobic interaction between the ligand and the receptor also represents good interaction.

The ADME studies of the synthesized compounds helped to conclude that all the compounds have good BBB penetration as that of the standard (data not shown). QPlogKp score indicates that all the test compounds and the standard are having skin permeability. QPlogKhsa score depicted all the compounds bound to the human serum albumin.

The 1,3,4-oxadiazole derivatives displayed moderate activity with bioactivity scores ranging from -0.01 to -0.05 against the properties of kinase and enzyme inhibition. The kinase inhibition of the derivatives indicates their significance towards antidepressant activity (data not shown).

LAZAR program was used to predict the toxicity of the synthesized compounds (data not shown). The compounds 4, 5, and 7 were liable to cause mutagenicity. Whereas compounds 4, 5, and 8 were liable to cause carcinogenicity, the remaining synthesized compounds were not found to be liable to cause any type of mutagenicity or carcinogenicity. All the synthesized compounds were not susceptible to cause acute toxicity and hence can be termed as safe.

## In vivo antidepressant activity

Forced swimming test results revealed that the compound **8** [4,4'-(1,3,4-oxadiazole-2,5-diyl) diphenol], **3** [3-(5-(4-hydroxyphenyl)-1,3,4-oxadiazol-2-yl) phenol] showed moderate activity when compared to the standard. The compound **1** [4-(5-(4-chlorophenyl)-1,3,4-oxadiazol-2-yl) phenol] also showed moderate antidepressant activity (Table III). Whereas, tail suspension test results revealed that compounds **8** and **1** showed moderate activity when compared to the standard.

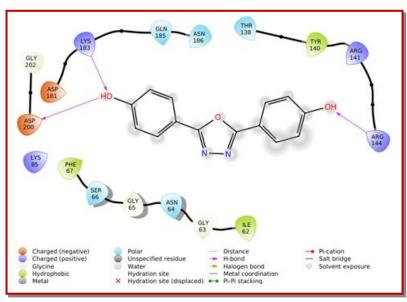


Figure 1: 2D interaction of compound 8 with 3GB2

Table III								
Antidepressant activity								
	Forced sw	imming test	Tail suspension test					
Compound	Duration of immobility (s)	%Change in immobility	Duration of immobility (s)	%Change in immobility				
1	150.2 ± 2.4 <sup>b</sup>	-30.0	147.0 ± 12.0 <sup>b</sup>	-36.6				
2	$183.6 \pm 1.7$ <sup>b</sup>	-14.5	$201.4 \pm 5.8$	-13.2				
3	$98.2 \pm 2.2^{b}$	-54.3	$148.4 \pm 14.5$ <sup>b</sup>	-36.0				
5	$175.2 \pm 1.9$ <sup>b</sup>	-18.4	$178.0 \pm 7.2^{a}$	-23.3				
8	$88.2 \pm 1.9$ <sup>b</sup>	-52.9	$106.0 \pm 20.0$ <sup>b</sup>	-54.3				
Control	$214.8 \pm 12.9$		$232.0 \pm 0.9$					
Standard (Imipramine)	75.4 ± 1.1	-64.9	$71.6 \pm 1.4$	-69.2				
Superscript "a" means p value <0.01; "b" means p value <0.0001								

Compound 3 showed moderate antidepressant activity.

#### Discussion

Among the synthesized 1,3,4-oxadiazole derivatives (1-8), five compounds (1, 2, 3, 5 and 8) were selected on the basis of the highest binding affinity. All these selected compounds a very high binding affinity to the selected GSK-3 $\beta$  receptor (PDBID: 3GB2). All the new com-pounds have significant interactions with the GSK-3 $\beta$  protein target. These five compounds screened for *in vivo* antidepressant activity by using the forced swimming test and tail suspension test models. These methods have been identified as making a situation where behavioral desperation is induced; that is, the animal loses hope of escaping the stressful environment that has been shown to play a role in major depression.

Imipramine and the tested compounds significantly reduced mice immobility time. Compounds 8, 3, 1 are the most efficient antidepressants, with immobility times that are identical to Imipramine. The molecule with the substitution of hydroxyl group (electron donating group) resulted in the strongest antidepressant efficacy in both test models (i.e. Compound 8). The recent literature also showed a promising activity against GSK-3β inhibitors (Tantray et al., 2018), and the same compounds were further evaluated as antidepressants. However the compounds were clubbed with 1,3,4-oxadiazole and benzimidazole moiety. Based on the putative role of GSK-3\beta activity (Tantray et al., 2016,), 1,3,4-oxadiazole-1,2,3-triazole conjugates were further investigated by antidepressants by adopting TST and FST methods. The later compounds were showed promising activity.

Substitution of hydroxyl group at position 4 and due to electron donating capacity showed significant activity, i.e., compound 8. Compounds with substitutions with electron withdrawing groups (chloro) at positions 4 showed moderate activity, i.e., compound 1. Compounds with substitutions with electron donating

groups (hydroxy) at positions 3 showed moderate activity, i.e., compound 3. Compounds with substitutions with electron donating groups (methoxy) at position 4 showed weak activity, i.e., compound 2.

Imipramine have been shown in numerous investigations to increase serotonin activity by decreasing biogenic amine reuptake in the CNS. In the present work, compound 8 and 3 showed highest affinity and binding energy with the GSK-3 $\beta$  target protein, resulting in highest antidepressants activity. The positive findings of this study imply that 1,3,4-oxadiazole derivatives synthesized as GSK-3 $\beta$  inhibitors could be employed as a therapeutic tool to analyze GSK-3 $\beta$  clinical implications and design viable therapeutics to treat these diseases.

## Conclusion

A new series of 1,3,4-oxadiazole derivatives were synthesized and evaluated for antidepressant activity. Some of the synthesized compounds (8) also showed significant *in vivo* antidepressant-like activity.

## **Financial Support**

Self-funded

## **Ethical Issue**

The study has been certified by the Ethical Committee for Animal Science (certificate No: NGSMIPS/IAEC/120). The treatment and handling of the animals were carried out in compliance with the guidelines defined by the Committee for the Management and Supervision of Animal Experiments.

### Conflict of Interest

Authors declare no conflict of interest

## Acknowledgement

The authors would like to thank Nitte CADD Lab, Nitte-Deemed to be University, Mangalore, India, for providing all the necessary facilities for carrying out this work and the University of Mysore, Institute of Excellence, Mysore, Karnataka for recording NMR and Mass spectral data.

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